

1. What force is responsible for holding the nucleus together? What particles does it act on? Does it work over long or short distances?  
**Strong Nuclear Force; It acts on all nucleons (protons & neutrons); short distances.**
2. What force acts on protons that are on opposite sides of the nucleus? Is it an attractive or a repulsive force?  
**Electric Force (Electrostatic); it is repulsive.**

The following questions refer to the “Band of Stability” diagram (see diagram).

3. What does each black dot represent?  
**A stable nucleus**
4. What is the most stable neutron/proton ratio for lighter elements?  
**1:1**
5. What is the most stable neutron/proton ratio for heavier elements?  
**1.5:1 (or 3:2)**
6. In the region above the band of stability, are there too many protons or neutrons?  
**neutrons ( $n/p > 1.5$ )**
7. In the region below the band of stability, are there too many protons or neutrons?  
**protons ( $n/p < 1$ )**
8. What two particles does a neutron decay into (one +, one -)? Write the decay equation that represents this. In what region does this decay occur?  
**a proton and an electron ( $\beta^-$  particle):  ${}_0^1\text{n} \rightarrow {}_1^1\text{p} + {}_{-1}^0\text{e}$ . It occurs above the band of stability.**
9. What two particles does a proton decay into (one +, one neutral)? Write the decay equation that represents this. In what region does this decay occur?  
**a positron ( $\beta^+$ ) and a neutron:  ${}_1^1\text{p} \rightarrow {}_0^1\text{n} + {}_{+1}^0\text{e}$ . It occurs below the band of stability.**
10. What is the primary decay mode for atoms with  $Z > 83$ ? What particle(s) are there too many of?  
 **$\alpha$ -decay. There are too many nucleons (protons & neutrons combined).**