

WKS 1-6: Naming & Formulas of Acids

In the table below, determine the name or the formula of the acid. From the formula, first determine the name of the anion, then determine the acid name. From the name, determine the anion formula first, then add enough H^+ ions to equalize charge

Formula to Name	Name to Formula
53. $HClO$	54. bromous acid
55. HF	56. hydroiodic acid
57. H_3PO_4	58. sulfuric acid
59. H_2CO_3	60. hydroselenic acid
61. $H_2C_2O_4$	62. acetic acid
63. H_2Te	64. chromic acid